

# Liquid Surface Spectrometer (LSS) at CMC



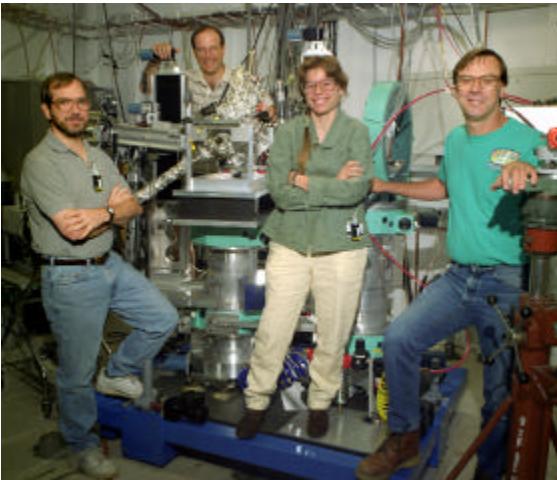
Ivan Kuzmenko

CMC

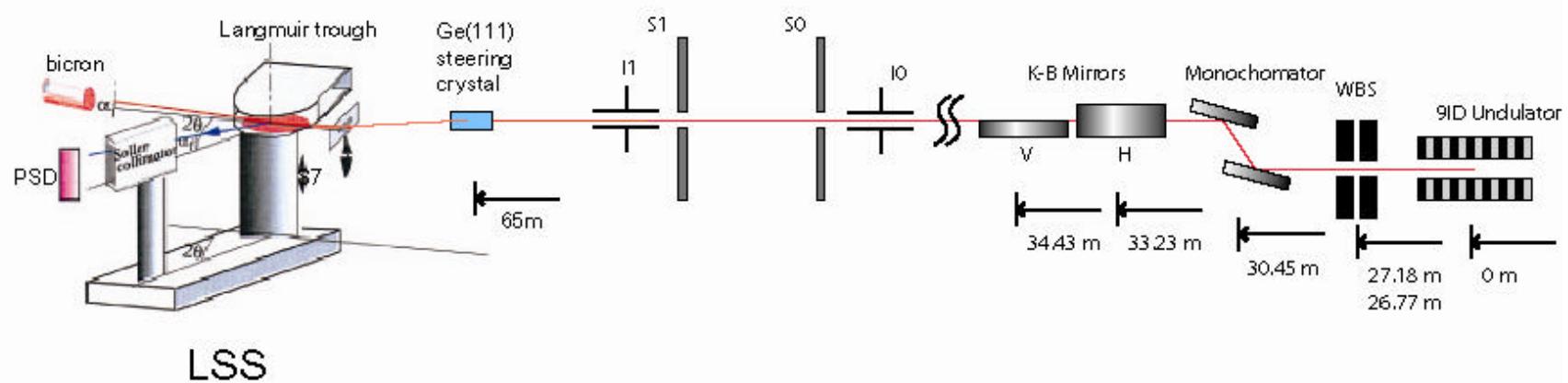
April 18, 2005

# Who is involved?

- People involved in design construction and commissioning of LSS (1997-2001):
  - Scott Coburn (BNL)
  - Ben Ocko (BNL)
  - Elaine DiMasi (BNL)
  - Thomas Gog (CMC)
  - Kent Blasie (UPENN)
- People currently involved in support and maintenance of LSS at 9-ID:
  - Ivan Kuzmenko (CMC)
  - Thomas Gog (CMC)
  - Diego Casa (CMC)



# Liquid Surface Spectrometer at 9-ID



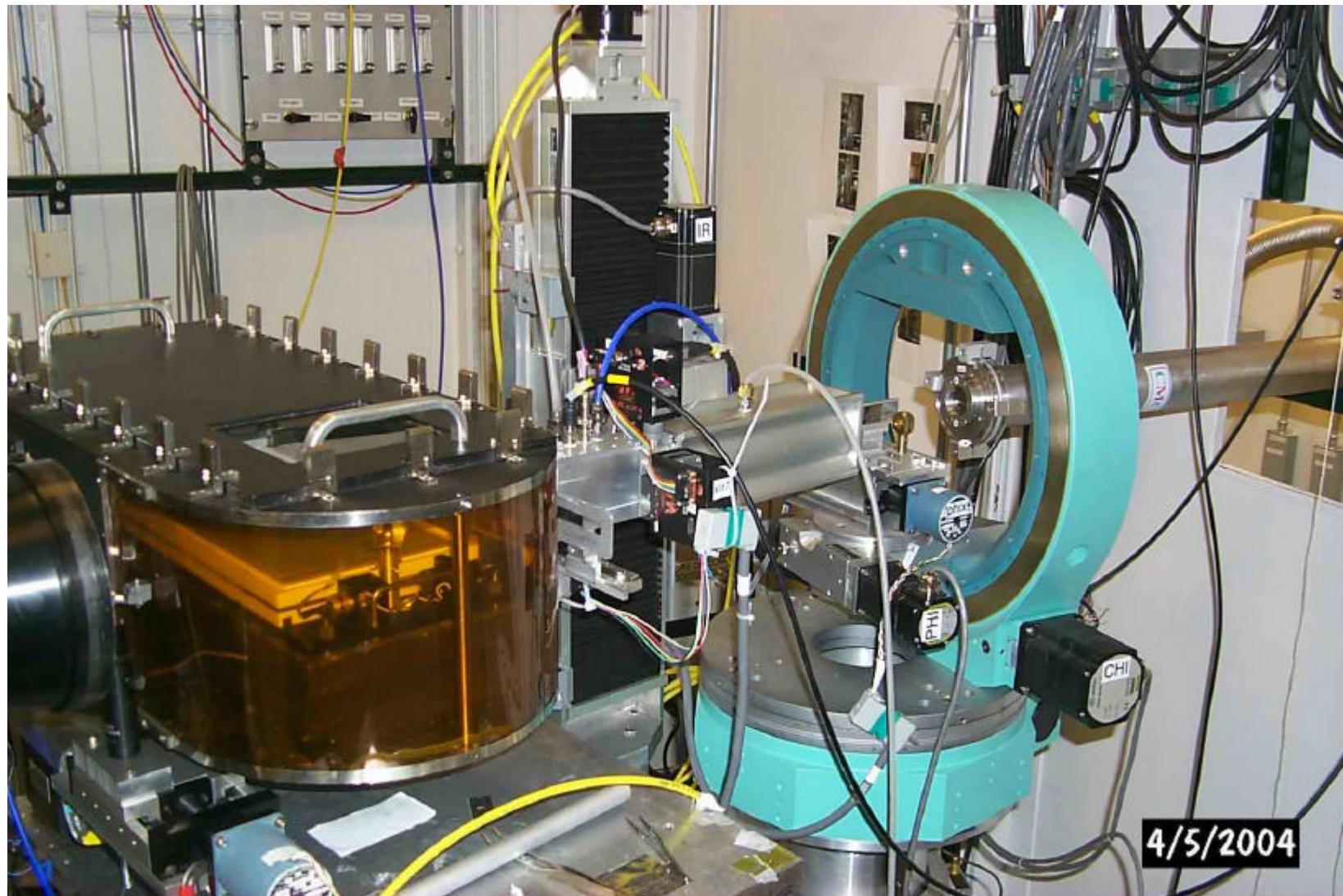
# Liquid Spectrometer Capabilities

- Reflectivity and Grazing Incident X-ray measurements (8-25KeV, typically)
  - Reflectivity measures surface normal density profile
  - GIXD measures the in-plane structure
- Large angular range
  - Required for liquid metal measurements
- Precise tracking,
  - required for ultra high resolution, analyzer, anomalous
- Two detector set-up (Bicron and PSD with soller slits)
  - Allows easy switch between different measurements
  - Heavy Sample chambers (UHV), weights up to 250 LB
- Very bright beam, well focused in vert. and horiz
- Optimized motion control for fast operations

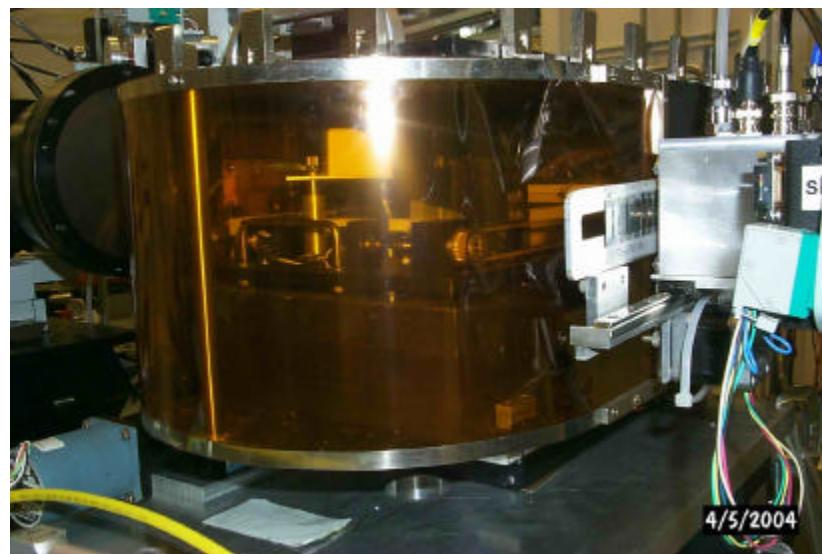
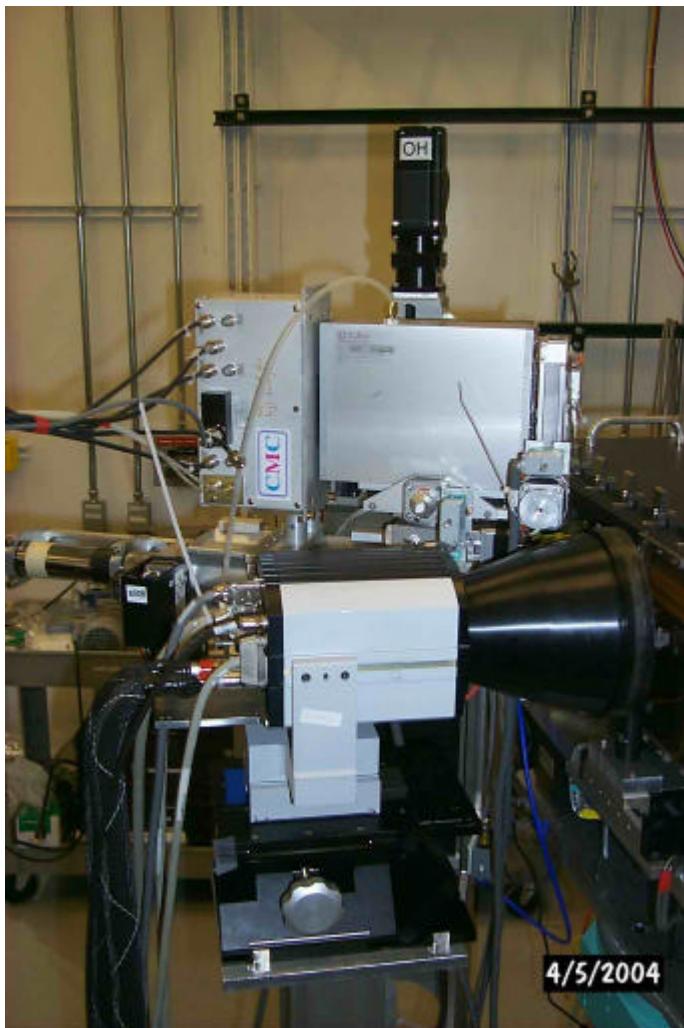
## Some Recent Features

- Si (111) asymmetric crystal was designed and commissioned for  $\lambda=1.3 \text{ \AA}$  ( $E=9.54\text{KeV}$ ) to defocus the incoming beam horizontally (10-fold) to avoid beam damage and improve sample statistics
- Ge (111) asymmetric crystal was designed and commissioned to overcome tracking problems at high energies ( $\sim 25\text{KeV}$ )
- Two dimensional detector (Bruker CCD) now can be mounted for in-plane GIXD measurements
- Oxygen sensor was installed to control the oxygen level in the trough

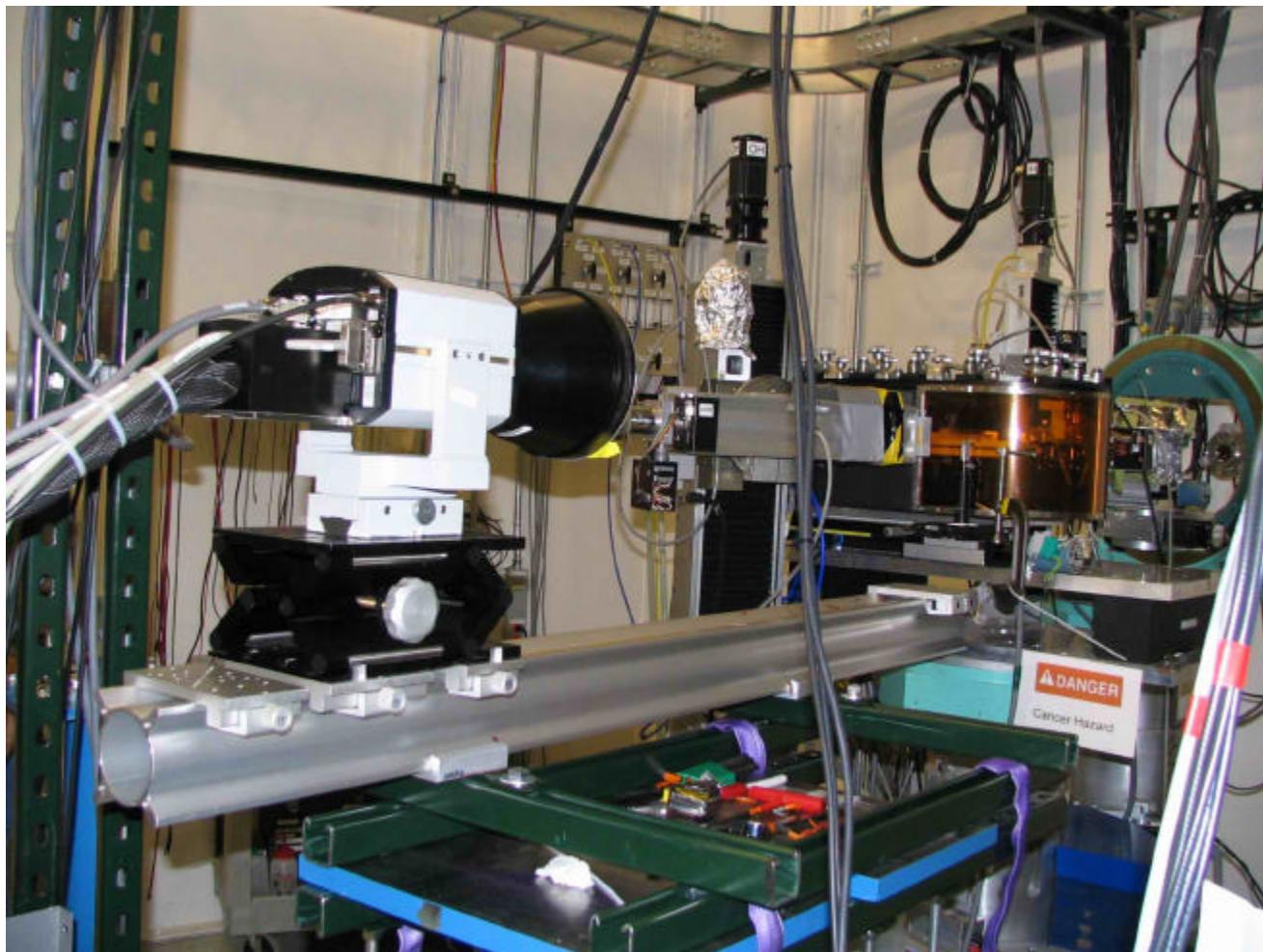
# LSS setup at CMC: pictures (1)



# LSS setup at CMC: pictures (2)



# LSS at CMC: some pictures (3)



# Scientific topics and publications (1)

- **Surface structure of Liquids:**

B. M. Ocko, Eric B. Sirota, M. Deutsch, E. DiMasi, S. Coburn, J. Strzalka, S. Zheng, A. Tronin, T. Gog, C. Venkataraman,  
Positional order and thermal expansion of surface crystalline N-alkane monolayers, Phys. Rev. E 63, 032602 (2001)

O. Shpyrko, M. Fukuto, P. Pershan, B. Ocko, I. Kuzmenko, T. Gog, M. Deutsch,  
Surface layering of liquids: The role of surface tension, Phys. Rev. B 69, 245423 (2004)

E. Sloutskin, B.M. Ocko, L. Tamam, I. Kuzmenko, T. Gog, and M. Deutsch,  
Surface Layering in Ionic Liquids: An X-ray Reflectivity Study (accepted for publication in JACS)

Brown M, Uran S, Law B, Marschand L, Lurio L, Kuzmenko I, Gog T,  
Ultra-stable oven designed for x-ray reflectometry and ellipsometry studies of liquid surfaces  
Rev. Sci. Ins. S 75 (8): 2536-2540 (2004)

L Marschand, M Brown, L B Lurio, B M Law, S Uran, I Kuzmenko, T Gog,  
X-Ray specular reflectivity study of scaling in critical binary fluid mixtures in the strong surface field limit  
(submitted for publication)

- **Geosciences:**

E. DiMasi, J. O. Fossum, T. Gog, C. T. Venkataraman,  
Orientational Order in Gravity Dispersed Clay Colloids: A Synchrotron X-ray Scattering Study of Na Fluorohectorite Suspensions, Phys. Rev. E 64, 061704 (2001)

- **Structure of buried interfaces:**

J Baumert, Michael Lefenfeld, E. Sloutskin, M. Deutsch, C. Nuckolls, B. Ocko  
Direct Observation of a Molecular Junction using High-Energy X-ray Reflectometry (work in progress)

# Scientific topics and publications (2)

- **Langmuir Films (Non Biological):**

Pao WJ, Zhang F, Heiney PA, Mitchell C, Cho WD, Percec V,  
[Grazing-incidence x-ray diffraction study of Langmuir films of amphiphilic monodendrons](#) Phys. Rev. E 67 (2), 021601 (2003)

Strzalka J, DiMasi E, Kuzmenko I, Gog T, Blasie JK,  
[Resonant x-ray reflectivity from a bromine-labeled fatty acid Langmuir monolayer](#)  
Phys. Rev. E 70 (5), 051603 (2004)

- **Langmuir Films (Biological):**

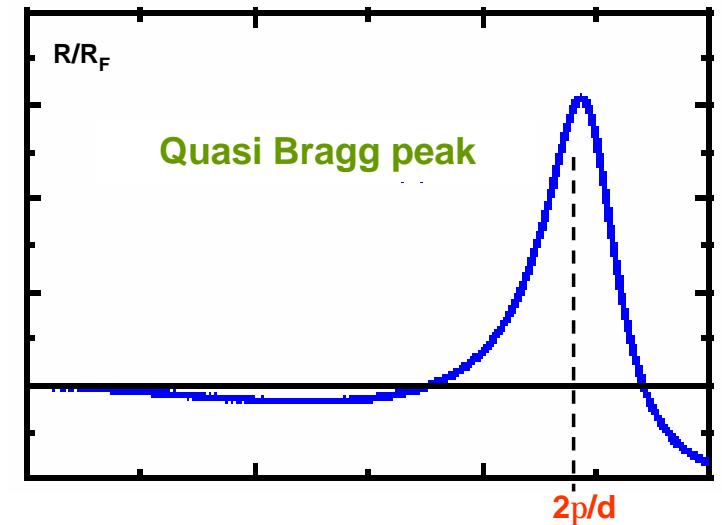
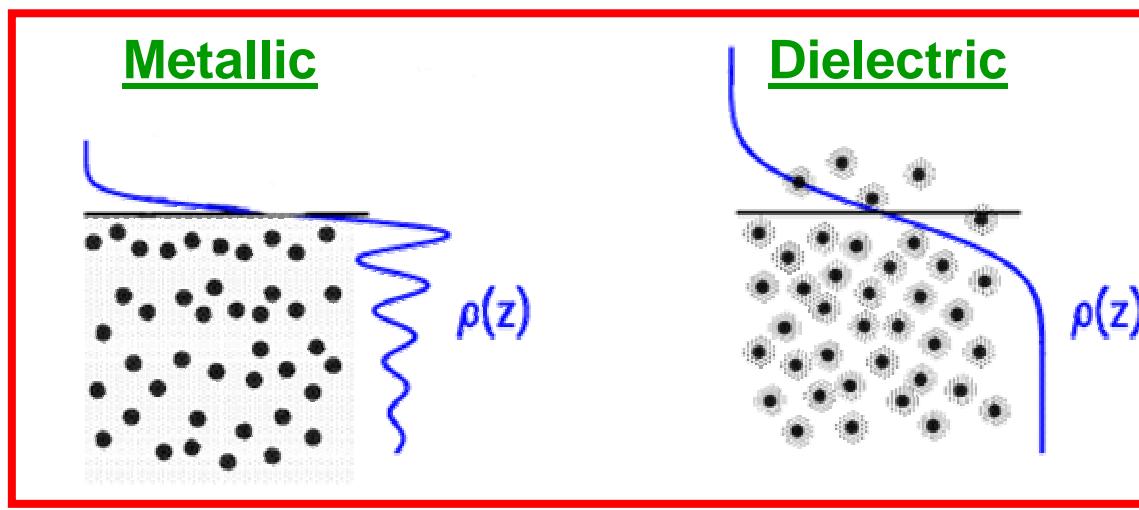
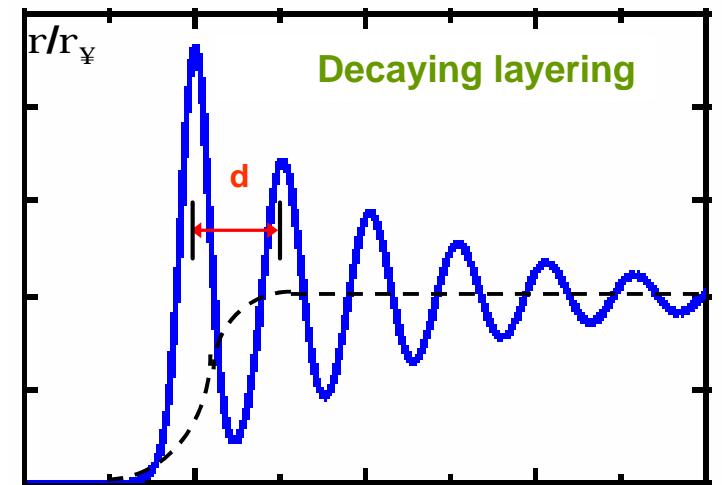
Zheng S, Strzalka J, Jones DH, Opella SJ, Blasie JK  
[Comparative structural studies of Vpu peptides in phospholipid monolayers by X-ray scattering](#) Biophys. J. 84 (4): 2393 (2003)

M. S. Kent, H. Yim, D. Y. Sasaki, S. Satija, J. Majewski, T. Gog,  
[Analysis of Myoglobin Adsorption to Cu\(II\)-IDA and Ni\(II\)-IDA Functionalized Langmuir Monolayers by Grazing Incidence Neutron and X-ray Techniques](#), Langmuir 20, 2819 (2004)

# Liquid metals vs dielectric liquids

S. Rice *et al.* (1960-75 ) and others:

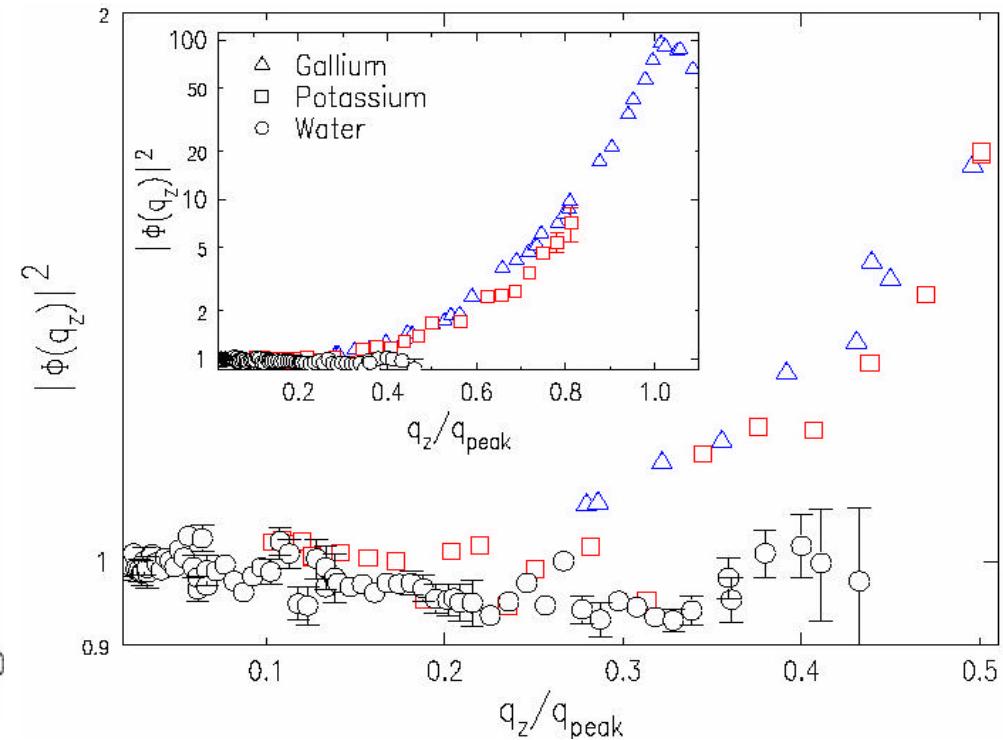
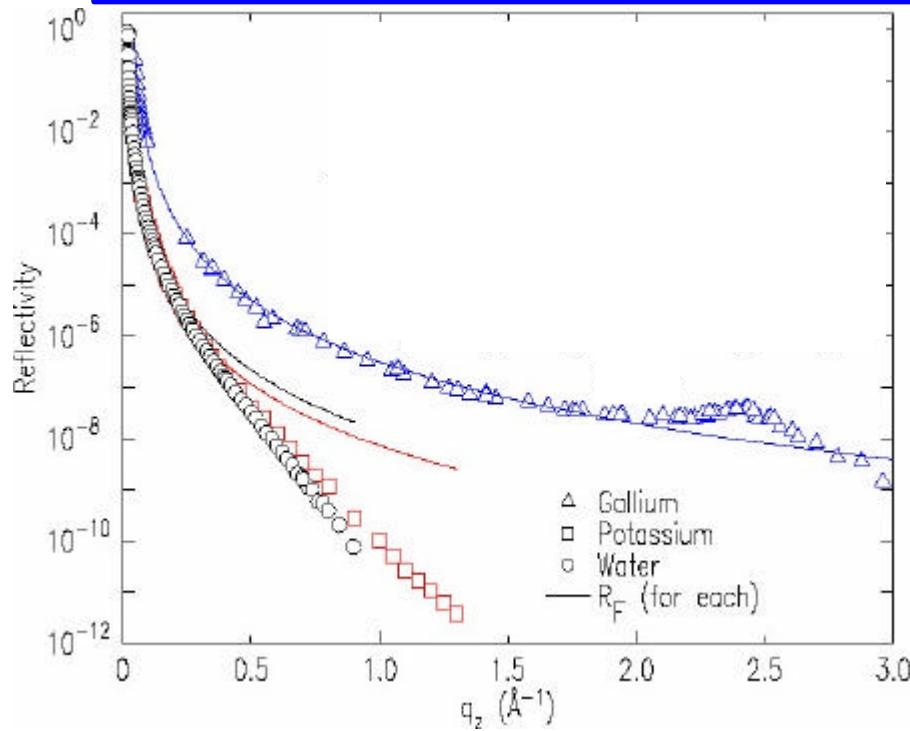
- Liquid metal: a strong variation in the interaction across the surface from conductor (bulk) to insulator (vapour).
- No such variation in a dielectric liquid, hence no layering should occur.



# Are all liquid surfaces intrinsically layered ?

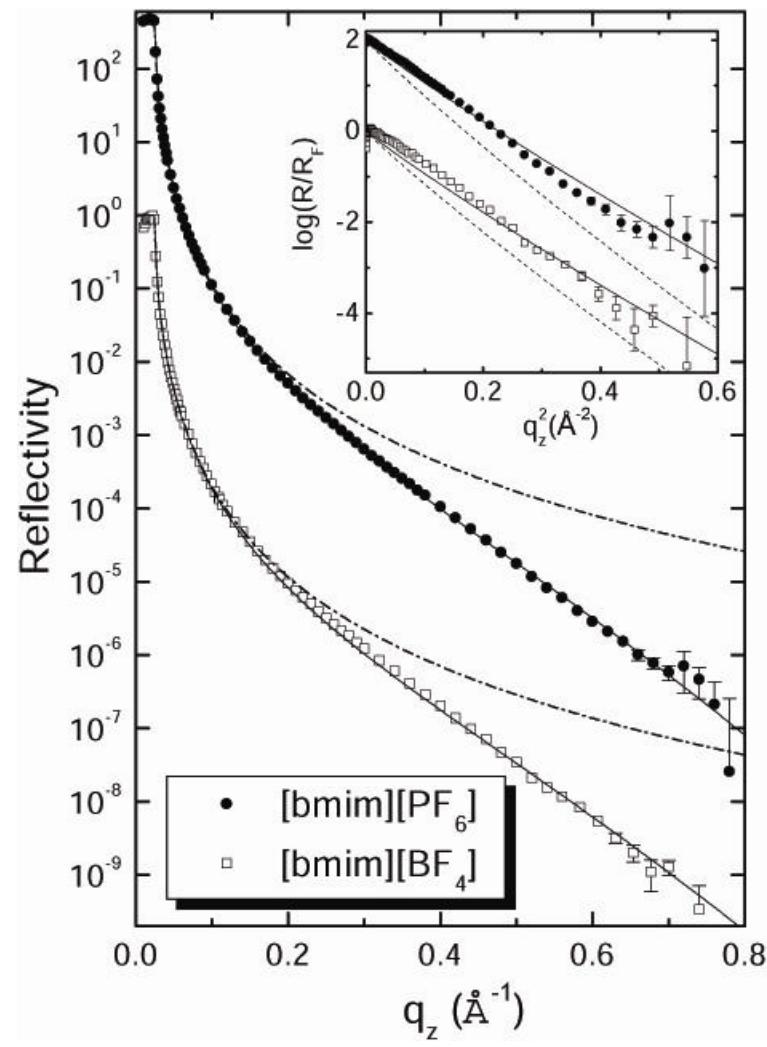
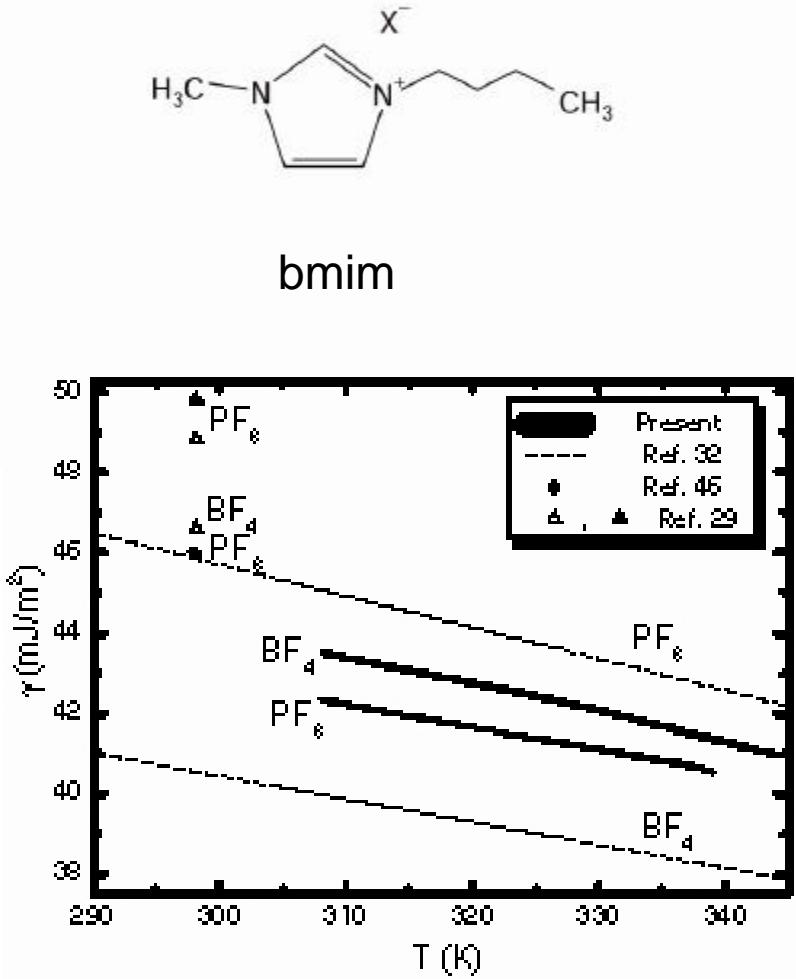
If K ( $g=110 \text{ mN/m}$ ) is layered, why not also water ( $g=72 \text{ mN/m}$ ) ?

Broader question: Is surface layering specific to liquid metals ?



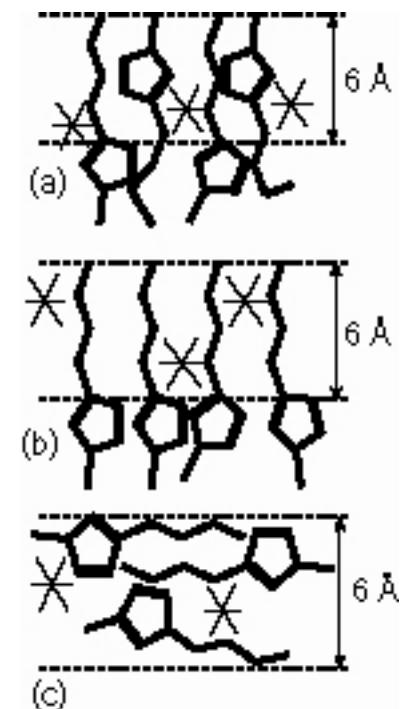
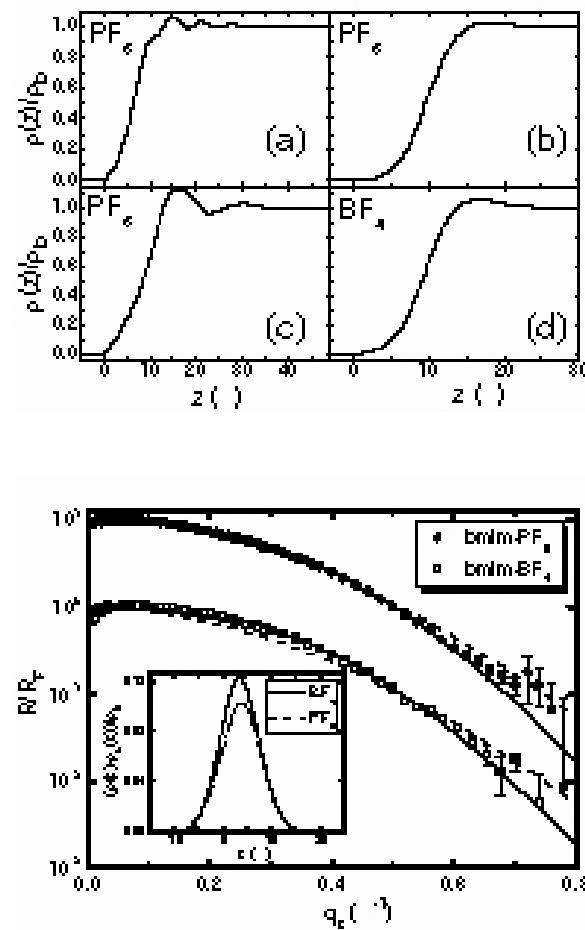
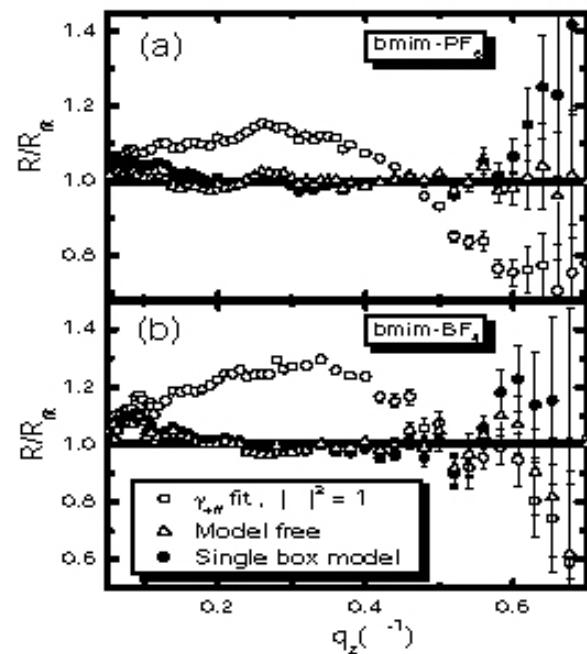
- Seems to support Rice et al. : layering is specific to liquid metals.
- Chacón et al. : all liquids layer for  $T/T_c \leq 0.2$ , if T is still above  $T_m$ .
- For water:  $T_m/T_c = 0.42$  → no layering expected and none indeed observed.
- For Ga:  $T_m/T_c = 0.043$  → layering is expected and indeed observed.
- What about propane, 1-butene, where  $T/T_c \leq 0.2$ ??

# Surface Lathering in Ionic Liquids (1)



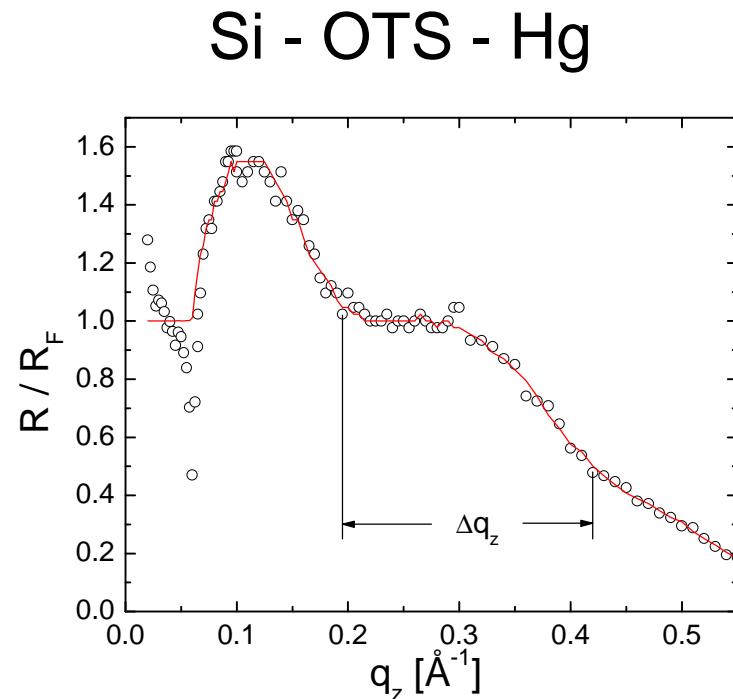
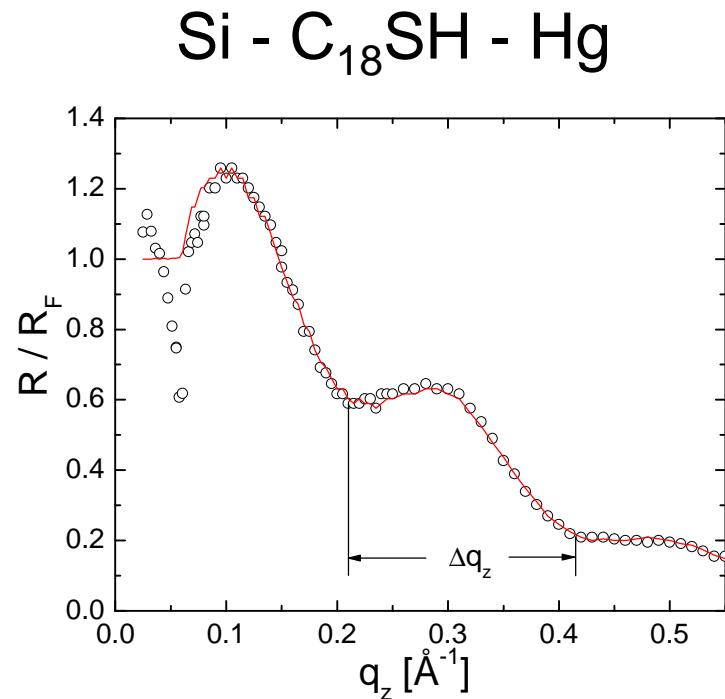
Slutskin et al (JACS)

# Surface Layering in Ionic Liquids (2)



Slutskin et al (JACS)

# Studies of buried interfaces (1)



→ “Kiessig” fringes clearly observable

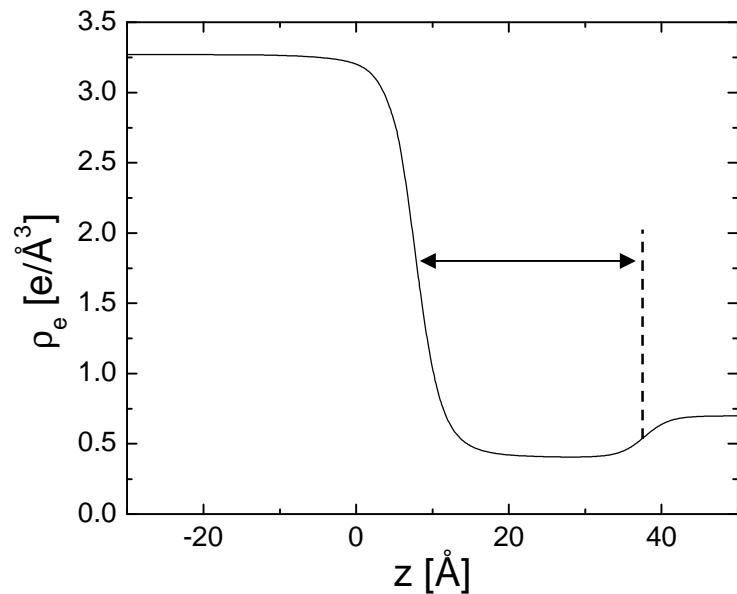
$$d \sim 29.9 \text{ \AA}$$

$$d \sim 27.3 \text{ \AA}$$

Julian Baumert et al (work in progress)

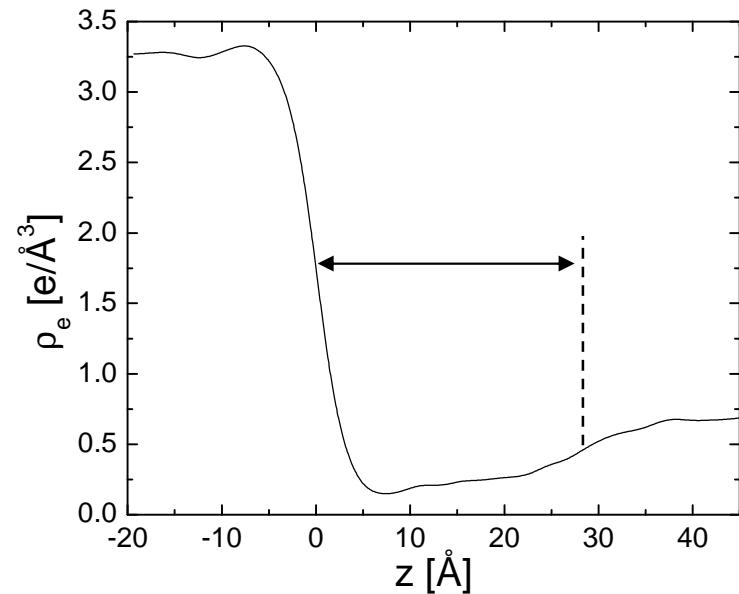
# Studies of buried interfaces (2)

Si - C<sub>18</sub>SH - Hg



- Electron density:  $\rho_e = 0.4 \text{ e/ \AA}^3$
- Thickness:  $d = 29.6 \text{ \AA}$

Si - OTS - Hg

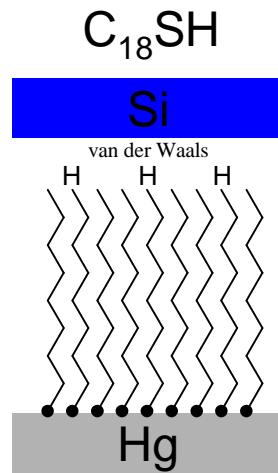


- Electron density:  $\rho_e = 0.24 \text{ e/ \AA}^3$
- Thickness:  $d = 28.4 \text{ \AA}$

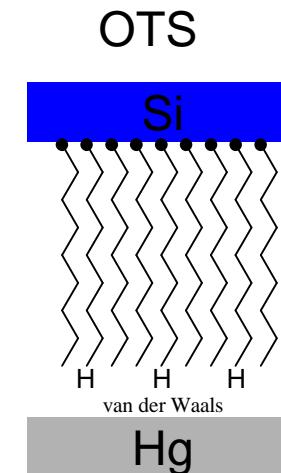
→ Densely packed monolayers in buried interfaces

# Structure of Buried Interfaces (3)

Si - C<sub>18</sub>SH - Hg: d = 29.6 Å  
Air - C<sub>18</sub>SH - Hg: l = 25.2 Å



Si - OTS - Hg: d = 28.4 Å  
Air - OTS - Si: l = 24.7 Å



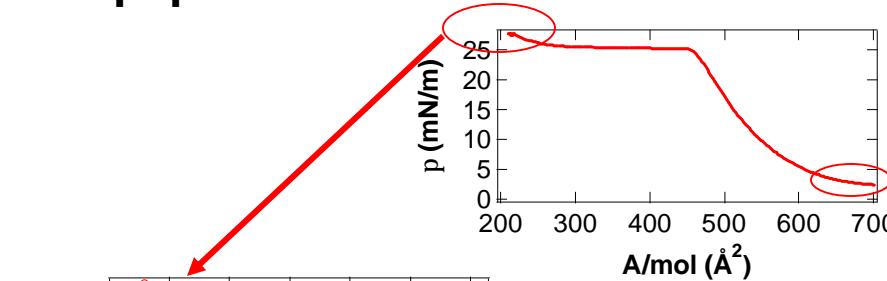
- C<sub>18</sub> – thiols and OTS molecular junctions are found to be 29.6 Å and 28.4 Å in length, consistently larger than monolayer thicknesses at the air interface.
  - The terminal hydrogen atom and the van der Waals radii of the hydrogen and the contacting surface (Thiol 4.3 Å, OTS 3.9 Å) account for the additional length.
- Thickness of Interface given by standing up molecules

# Resonance Reflectivity (1)

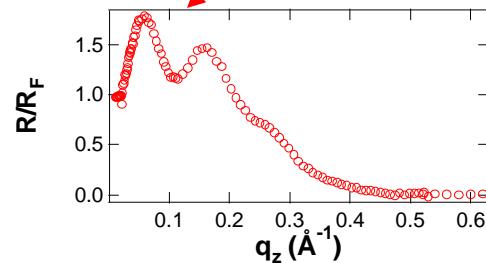
X-Ray Reflectivity measures the profile structure  $\rho(z)$  of Langmuir Monolayers

$\rho$  shows  $\alpha$ -helical peptide bundles can be oriented with surface pressure,  $p$

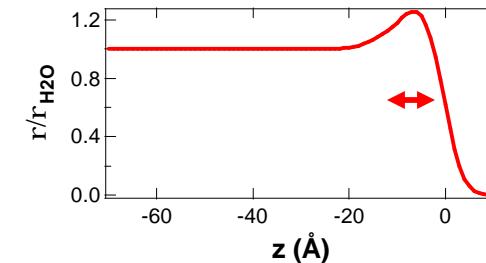
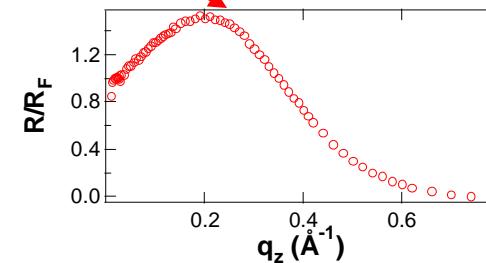
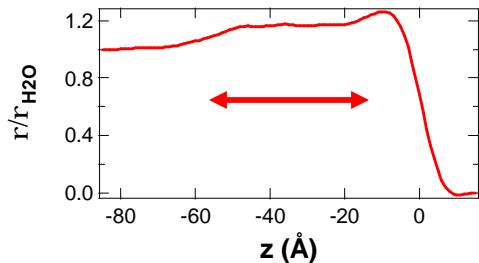
Isotherm



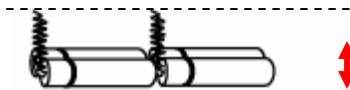
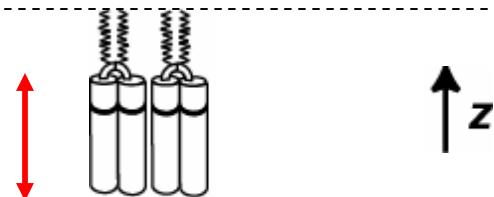
XR data



Profiles:  
 $\rho(z)$



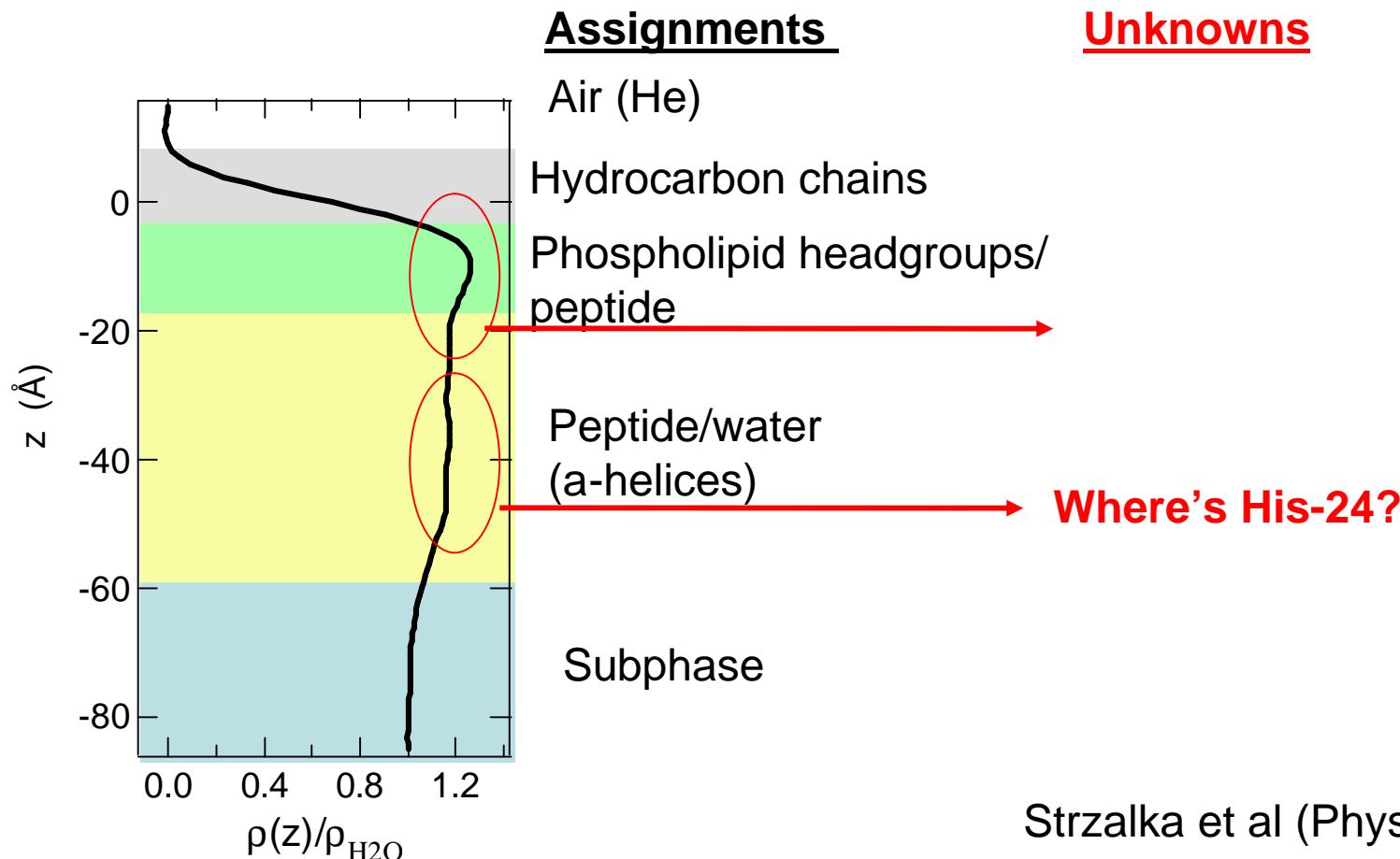
Interpretation



Strzalka et al (Phys.Rev.E)

# Resonance Reflectivity (2)

X-Ray Reflectivity identifies regions, shapes, not specific groups



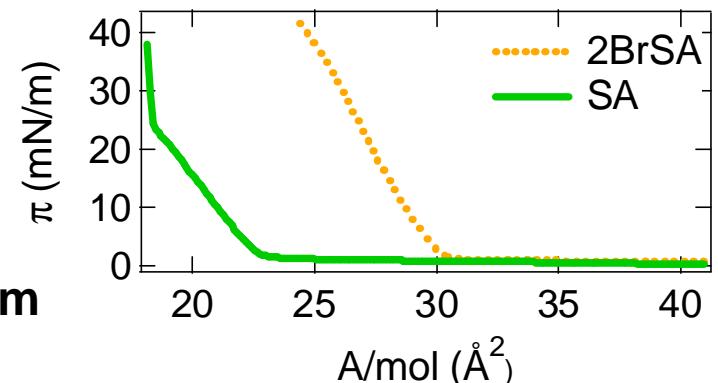
Strzalka et al (Phys.Rev.E)

# Resonant X-ray Reflectivity (3): Fatty Acid Methods

Samples: **2-Bromo-Stearic Acid (2BrSA)** and **Stearic Acid (SA)**  
 $\text{CH}_3(\text{CH}_2)_{15}\text{CBrHCOOH}$   
Br-labeled  
 $\text{CH}_3(\text{CH}_2)_{16}\text{COOH}$   
**CONTROL**

Measurements: **13 474 eV (Br K edge)**  
**13 674 eV (Br K+d)**  
**13 274 eV (Br K-d)**  
step sample 40 $\mu\text{m}$  between points

Monolayers: pure **2BrSA** or **SA** at  $p=5 \text{ mN/m}$   
MilliQ water, 5°C

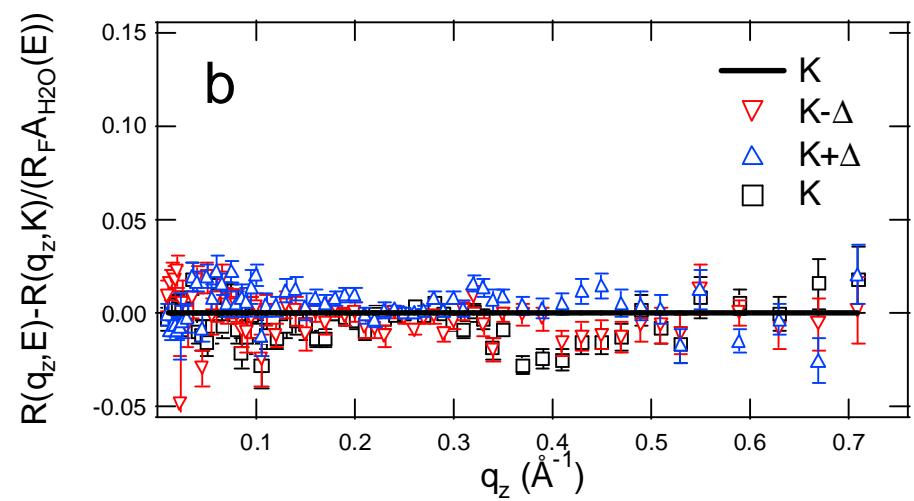
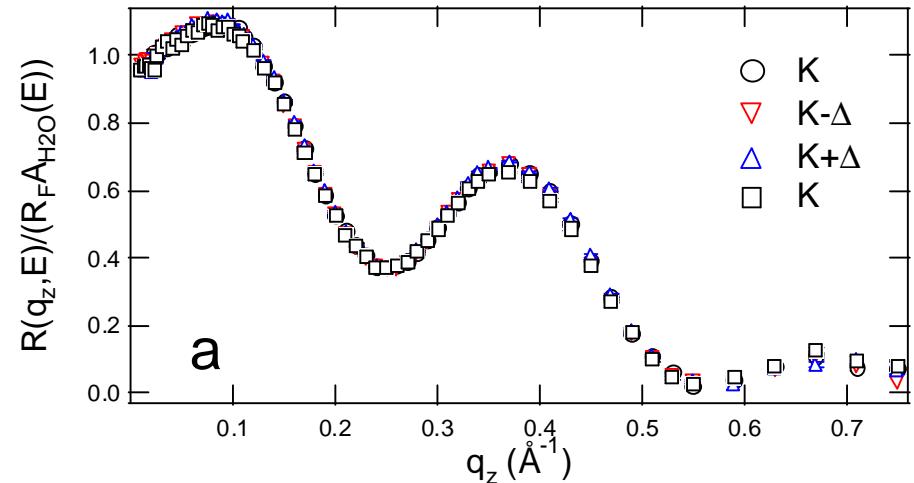
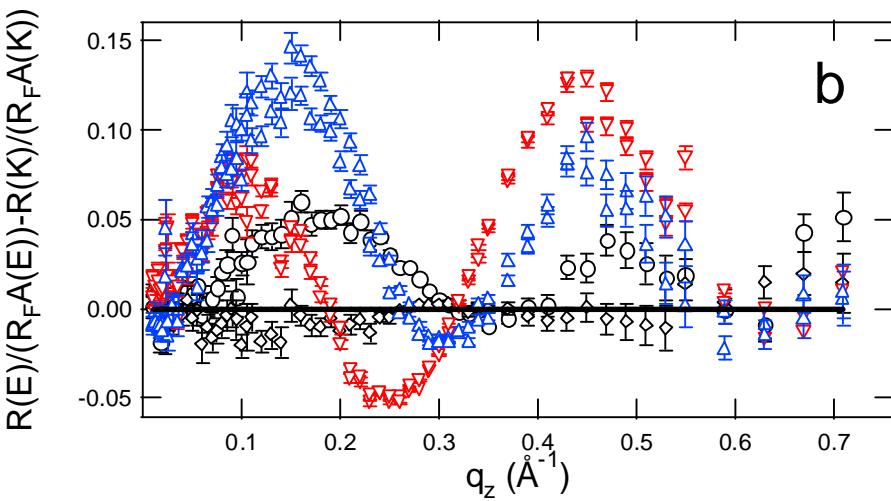
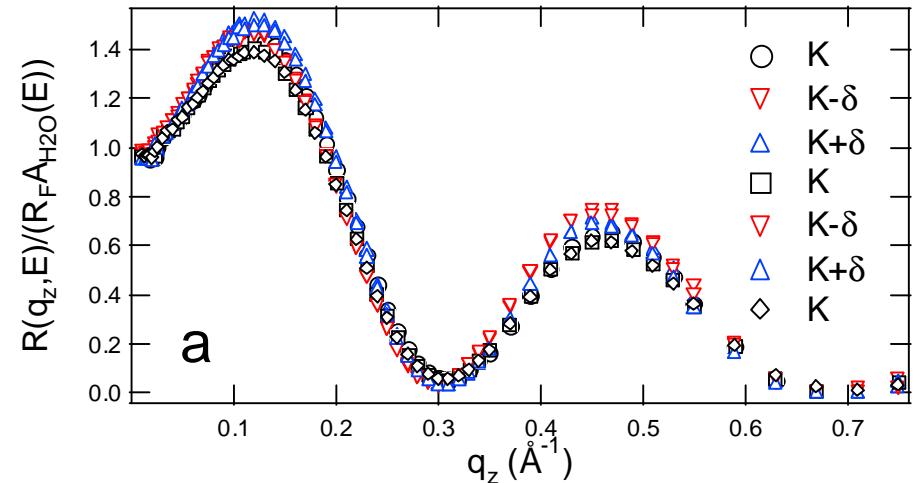


X-rays: Liquid Surface Spectrometer at CMC CAT,  
Sector 9 of Advanced Photon Source (Argonne, IL)

# Resonant X-ray Reflectivity (4): Fatty Acid

**2-bromo-stearic acid (2BrSA)**

**Control: stearic acid (SA)**



Strzalka et al (Phys.Rev.E)